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compare acids and bases

and understand why their

strengths vary. You will

define pH and pOH and

calculate the pH and pOH

of aqueous solutions. You

will calculate acid and base concentrations and determine concentrations experimentally. You will explain how buffers resist changes in pH. Chapter 19: Acids and Bases Chapter 19... Acids, Bases, and Salts ACID-BASE THEORIES Acids and bases are all around us and part of our everyday life (ex. bodily functions, vinegar, carbonated drinks, citrus fruits, car batteries, antacids, cleaning products, etc.). Acids and bases have the following properties: Acids Bases Tart or sour taste

Bitter taste Chapter 19 Acids, Bases, and Salts - Science with Mrs ... Section 19.1 - Acid-Base Theories Acids have a sour taste, change the color of an indicator, can be strong or weak electrolytes in aqueous solution, and react with metals. Bases Chapter 19 - Acids, Bases, and Salts Chapter 19 "Acids, Bases, and Salts ... Figure 19.22, page 615. Neutralization. The solution of known concentration is called the standard solution. added by using a buret. Continue

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